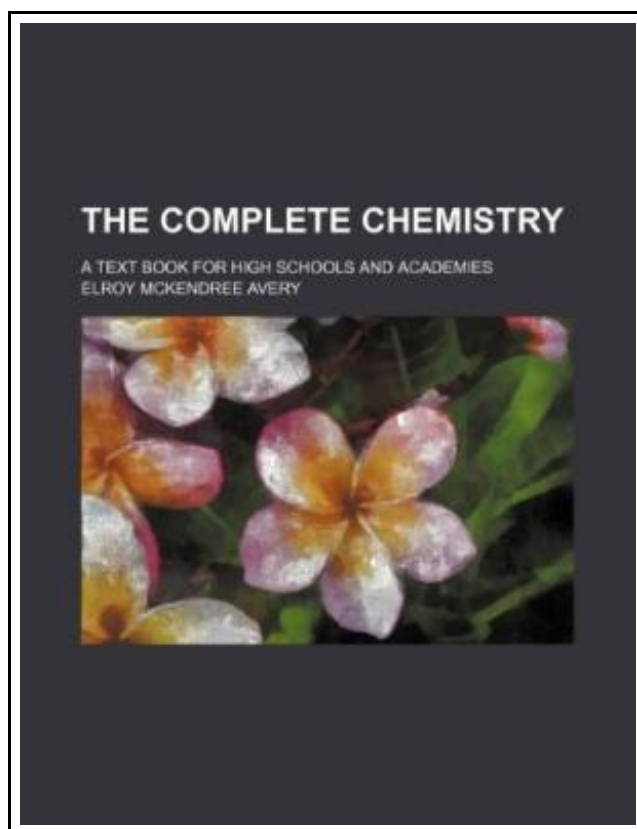


# The Complete Chemistry; A Text Book for High Schools and Academies



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## ***Reviews***

*If you need to adding benefit, a must buy book. it absolutely was writtern extremely flawlessly and valuable. Once you begin to read the book, it is extremely difficult to leave it before concluding.  
(Mrs. Odie Murphy II)*

## THE COMPLETE CHEMISTRY; A TEXT BOOK FOR HIGH SCHOOLS AND ACADEMIES



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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book \*\*\*\*\* Print on Demand \*\*\*\*\*.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1883 Excerpt: . the apparent quantivalence of P in  $\text{H}_2\text{PO}_3$ ? (6.) In  $\text{H}_2\text{PO}_4$ ? 28. Give the name of the compound graphically symbolized as follows:  $(\text{HO})\text{--P--O--}(\text{HO})$  0  $(\text{HO})\text{--P--O--}(\text{HO})$  29. What does the above graphic symbol intimate concerning the quantivalence of the P? 30. Give the empirical symbol for the compound graphically symbolized as follows:  $(\text{HO})\text{--p}$  0  $(\text{HOK})\text{--}(\text{HO})\text{p}$  (  $(\text{HOKK})$  31. What does this last graphic symbol intimate concerning the quantivalence of the P 1 32. Write the symbol for hypophosphorous acid upon the ammonia type. 33. What weight of P in 10 l. of  $\text{PH}_2$  under normal conditions? 34. (a.) How many ev. cm. of  $\text{NH}_2$  can be obtained from 53.5 g. of  $\text{NH}_4\text{Cl}$ ? (6.) How many, under the conditions, 15C. and 740 mm. (Ph., 494.) JCTTON IIL ARSENIC AND ITS COMPOUNDS. Symbol, As; specific gravity, 5.6 to 5.9; atomic weight, 75 m. c.; molecular weight, 300 m. c.; quantivalence, 3 or 5. 243. Occurrence and Preparation.--Arsenic is widely distributed in small quantities. It is sometimes found free in nature but more frequently combined with iron, sulphur and other elements. It is generally prepared from arsenical pyrite (mispickel,  $\text{FeSAs}$ ) by sublimation or from its oxide by reduction with charcoal. 244. Properties and Uses.--Arsenic has a metallic lustre and a steel-gray color. Its vapor is 150 times as heavy as hydrogen. In its physical properties, it closely resembles a metal; in its chemical properties, it more closely resembles a non-metal. It has been called the connecting link between the...



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